

Summary

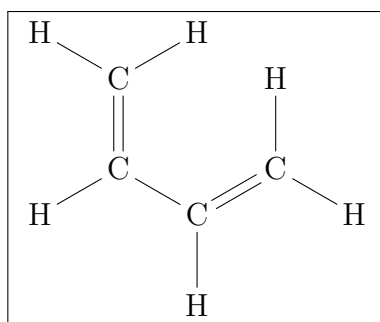
August 28, 2026

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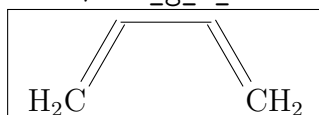
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0.1 Loaded Graphs

0.1.1 Butadien



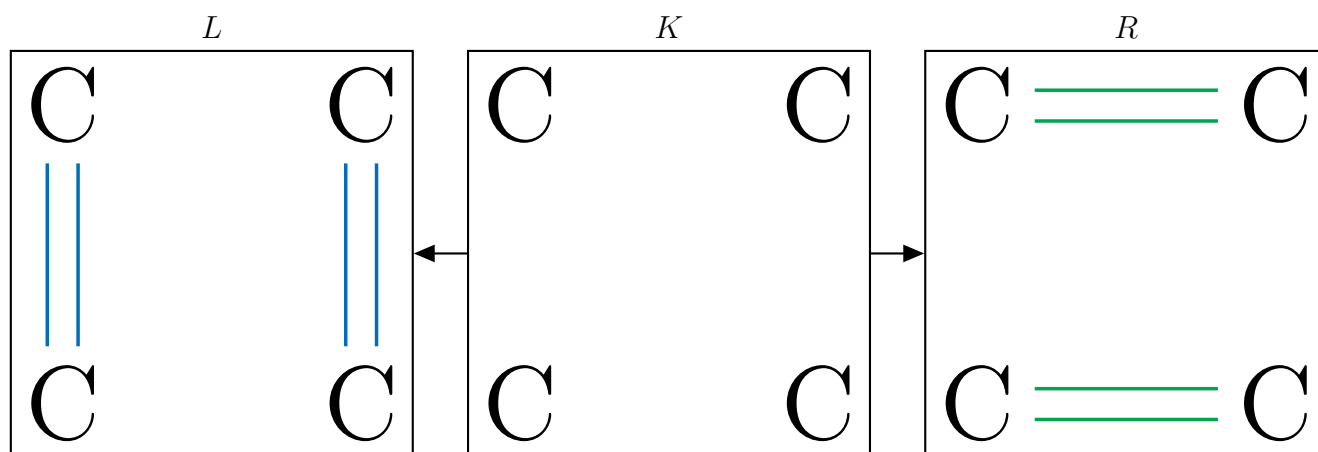
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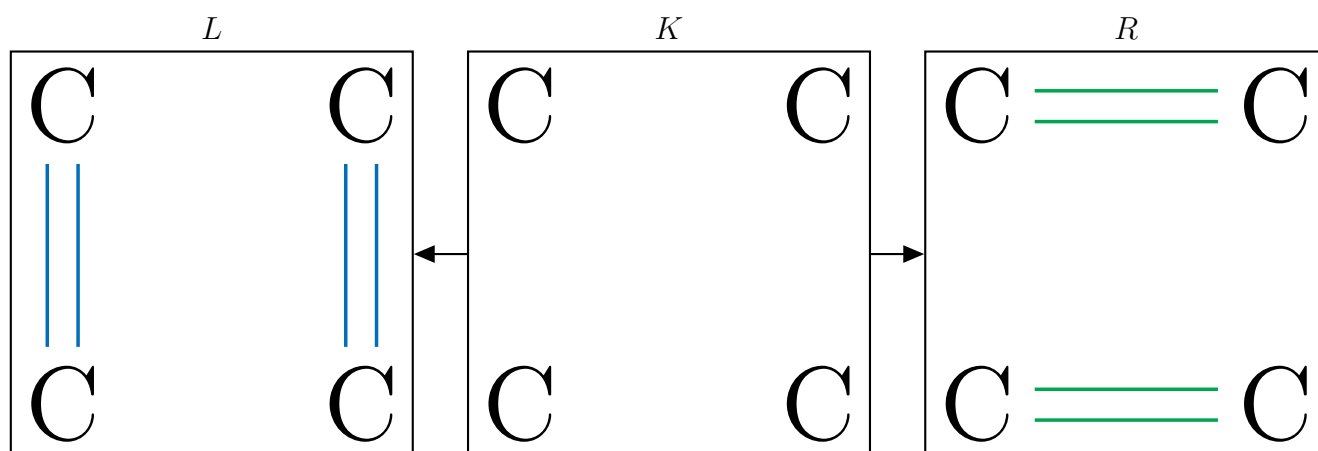
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0.2 Loaded Rules

0.2.1 $r_{\{0\}}$

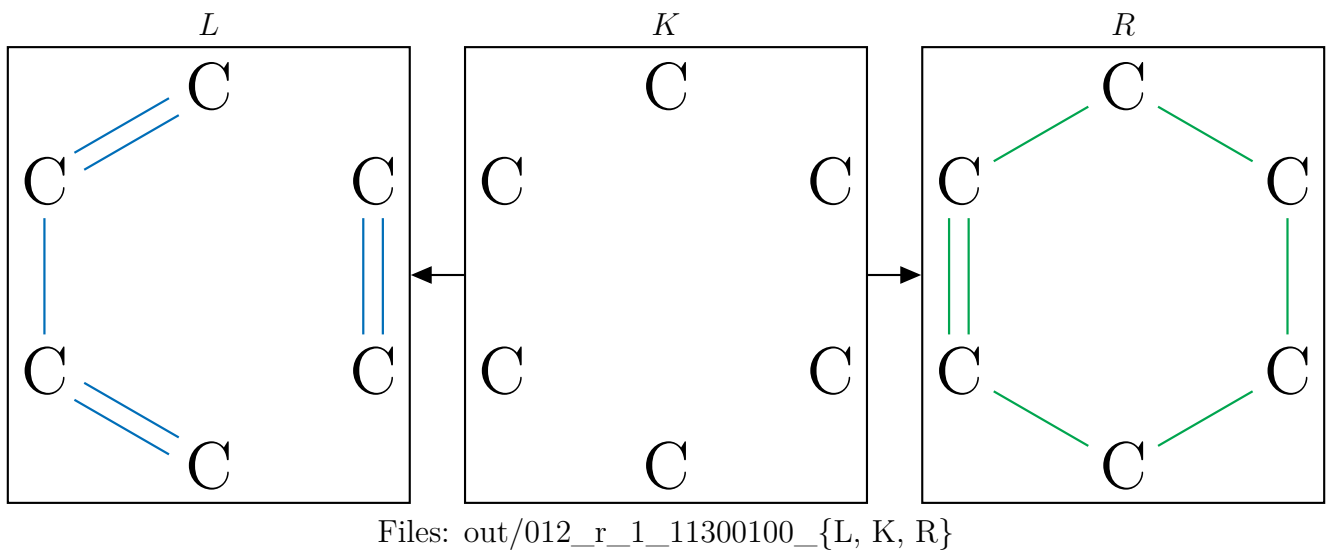
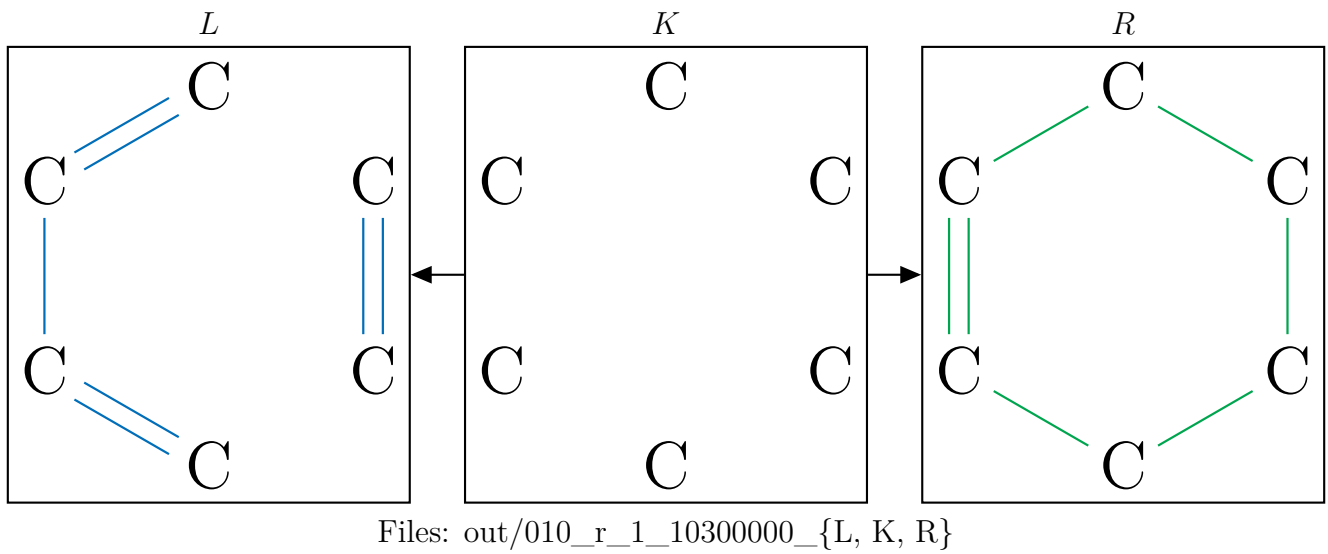


Files: out/005_r_0_10300000_{L, K, R}



Files: out/007_r_0_11300100_{L, K, R}

0.2.2 $r_{\{1\}}$



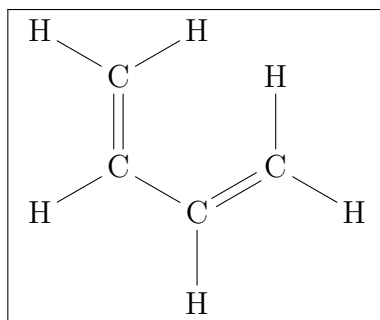
0.2.3 DG Hyper, dg_0

Figure too large, see out/014_dg_0_11100_coord.pdf

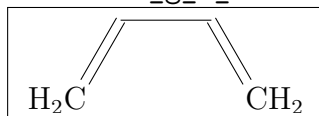
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0.3 Product Graphs

0.3.1 Butadien

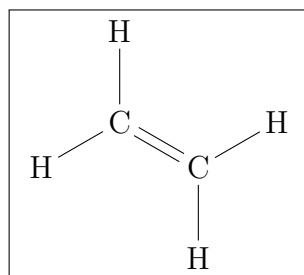


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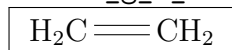


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0.3.2 $p_{\{0,0\}}$

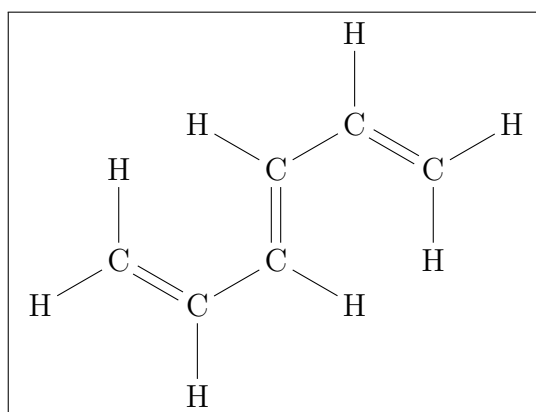


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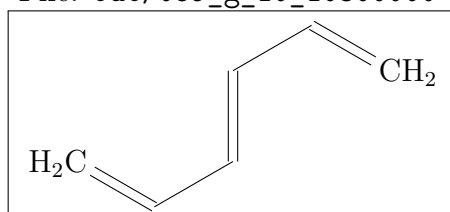


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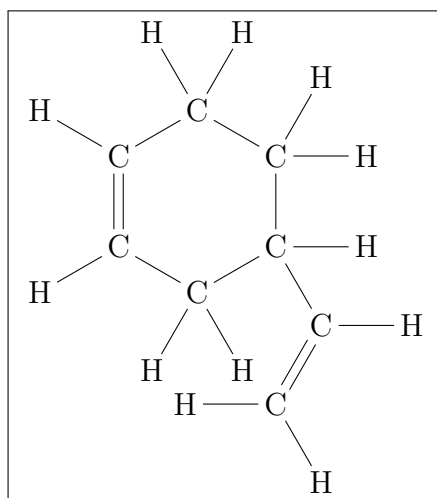
0.3.3 $p_{\{0,1\}}$



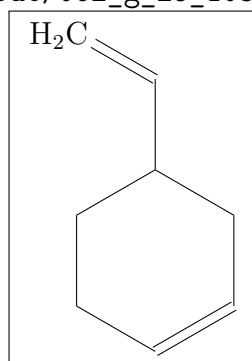
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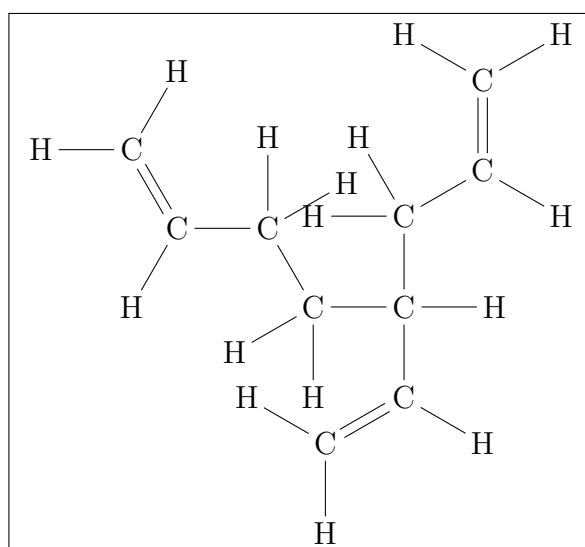


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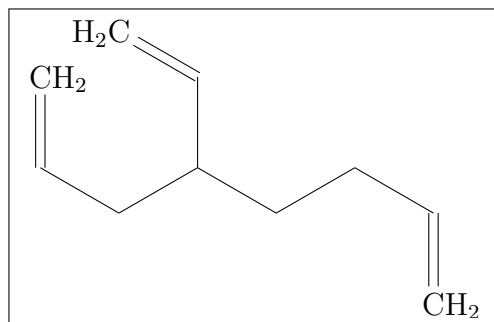


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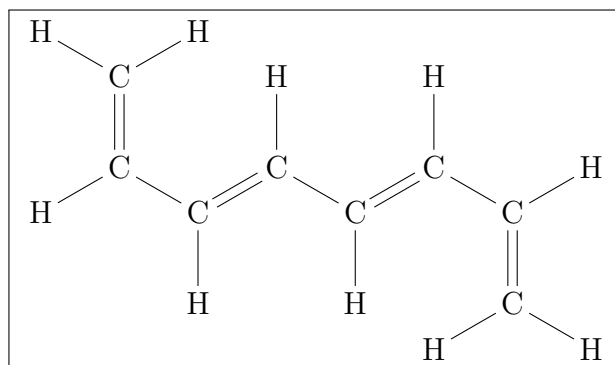


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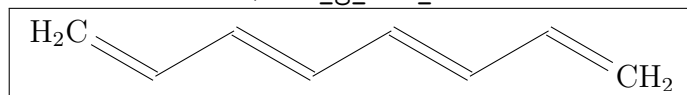


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0.3.6 $p_{\{0,4\}}$

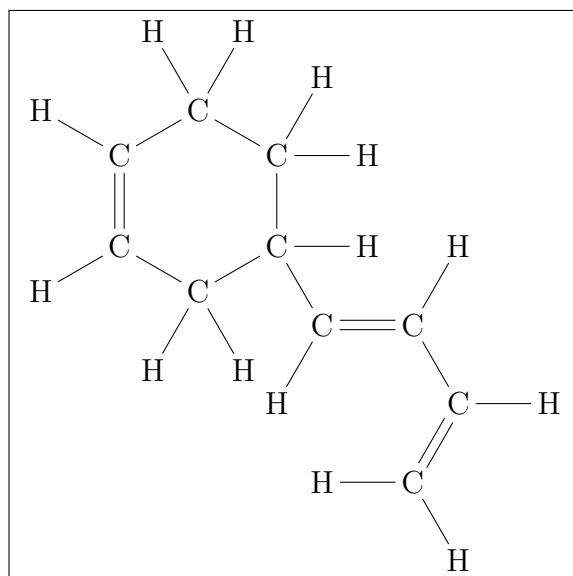


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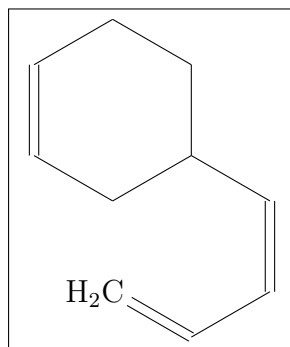


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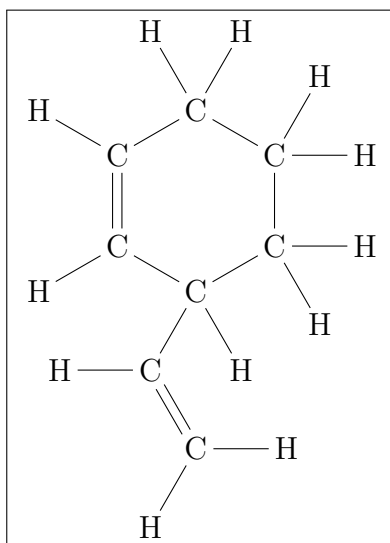


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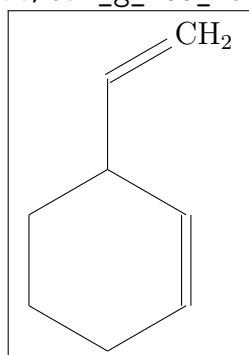


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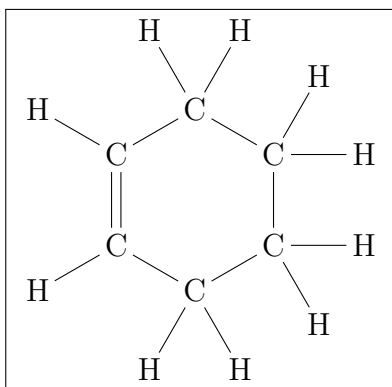


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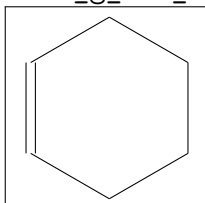


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0.3.9 $p_{\{0,7\}}$

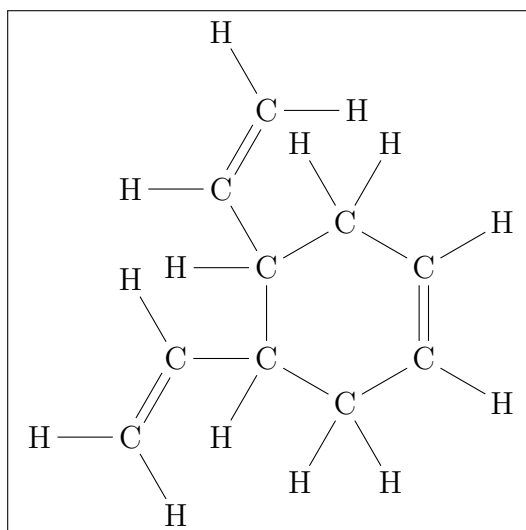


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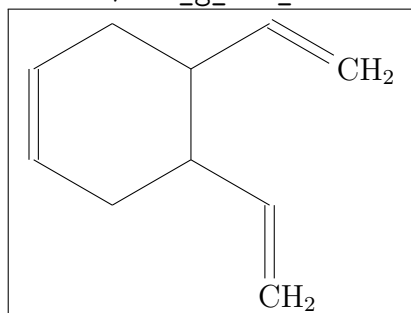


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0.3.10 $p_{\{0,8\}}$

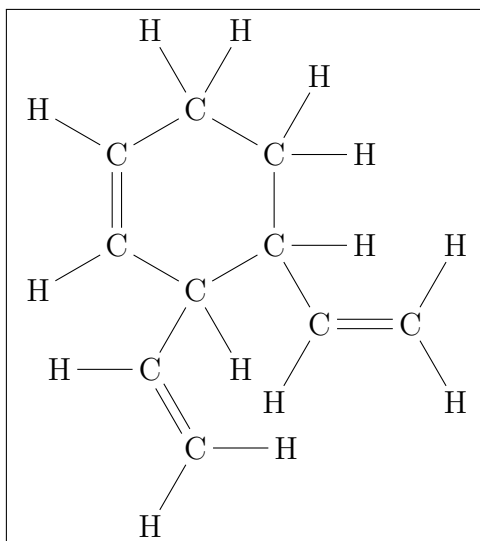


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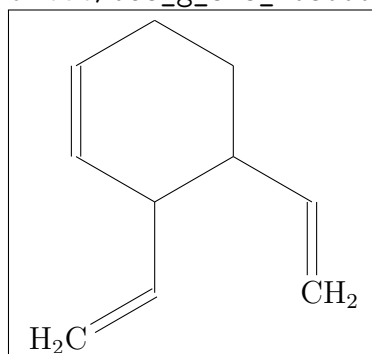


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0.3.11 p_{0,9}

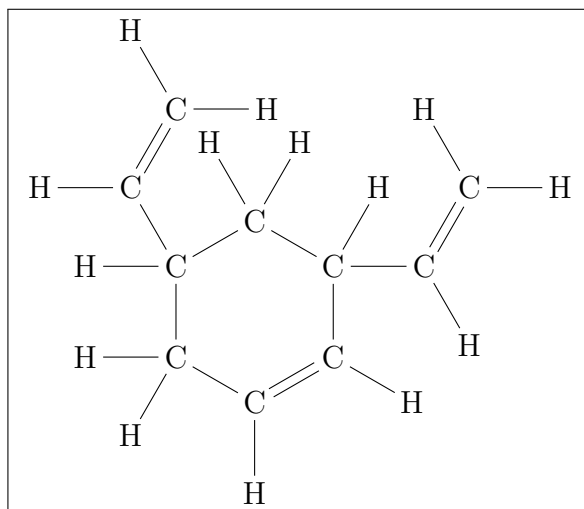


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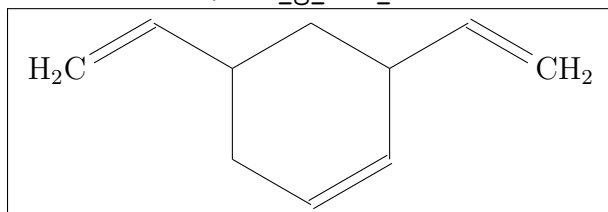


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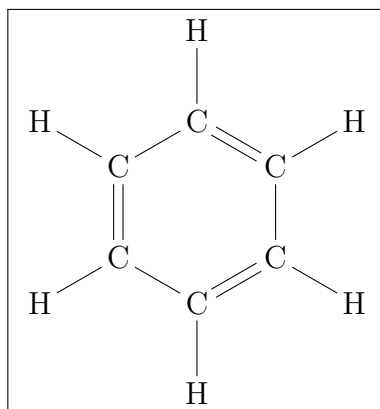


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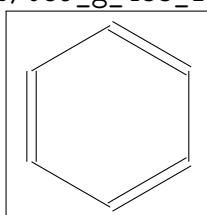


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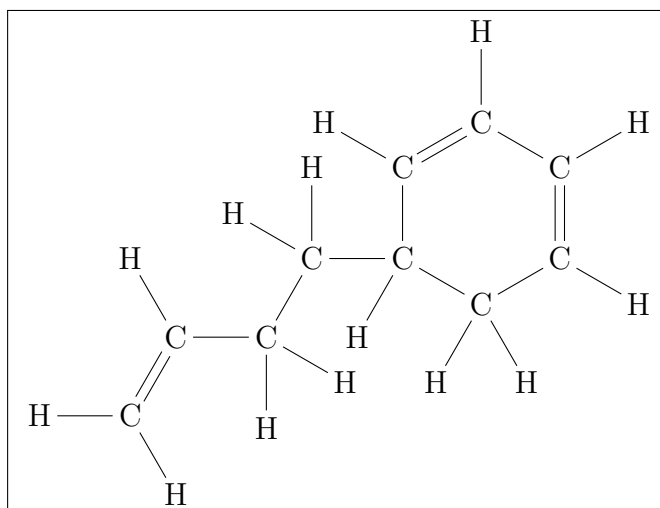


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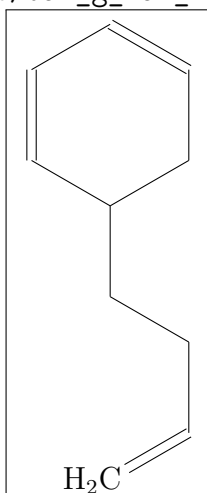


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0.3.14 $p_{\{0,12\}}$

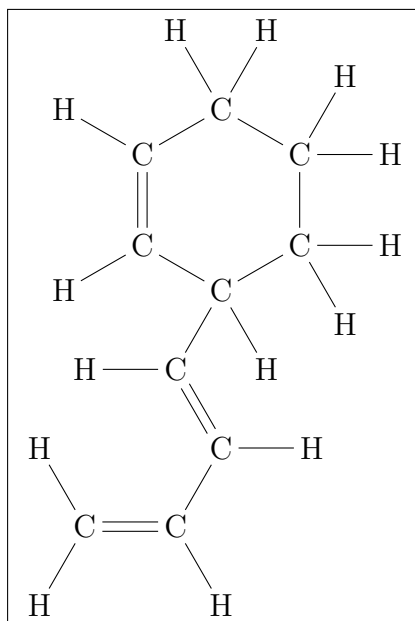


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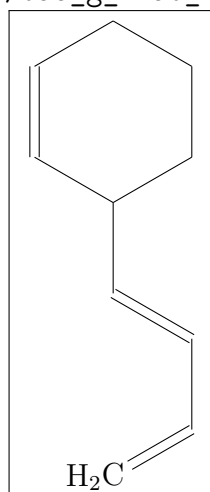


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0.3.15 p_{0,13}

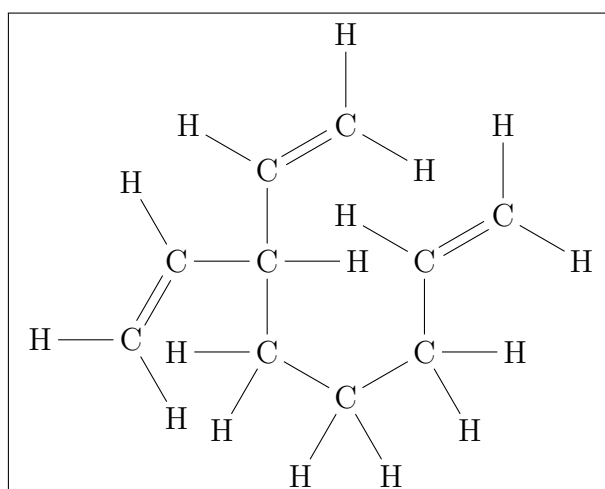


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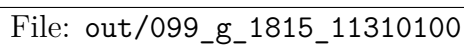


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0.3.16 p_{0,14}



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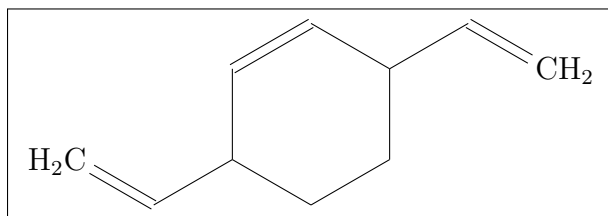


The diagram shows the chemical structure of 1,2,3,4,5,6-hexachlorocyclohexa-1,3,5-triene, commonly known as Dieldrin. It consists of a six-membered carbon ring with three alternating double bonds. Each carbon atom is bonded to one hydrogen atom and one chlorine atom. The chlorine atoms are positioned at the 1, 2, 3, 4, 5, and 6 positions, making the molecule highly toxic and persistent in the environment.

C=CCCCC=C

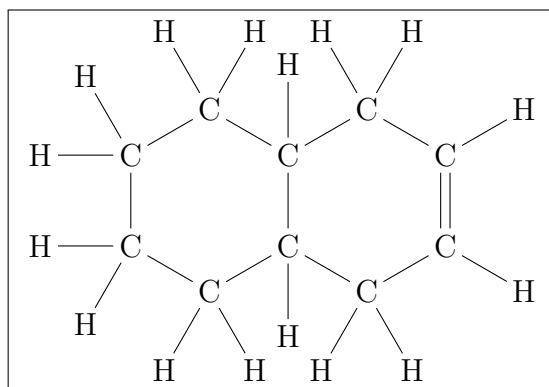
The diagram shows the chemical structure of 1,3-cyclohexadiene, a six-membered carbon ring with two double bonds. The carbons are arranged in a hexagonal pattern. Starting from the top and moving clockwise, the first carbon is part of a double bond with the second carbon and has one hydrogen atom attached above it. The second carbon is part of a double bond with the first carbon and has one hydrogen atom attached to its right. The third carbon is single-bonded to the second and fourth carbons and has two hydrogen atoms attached (one above, one to the right). The fourth carbon is single-bonded to the third and fifth carbons and has one hydrogen atom attached to its right. The fifth carbon is part of a double bond with the sixth carbon and has one hydrogen atom attached below it. The sixth carbon is part of a double bond with the fifth carbon and has one hydrogen atom attached to its right. The seventh carbon is single-bonded to the sixth and first carbons and has two hydrogen atoms attached (one to the left, one below).

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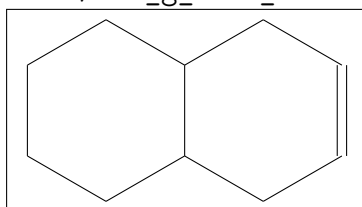


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0.3.19 $p_{\{0,17\}}$

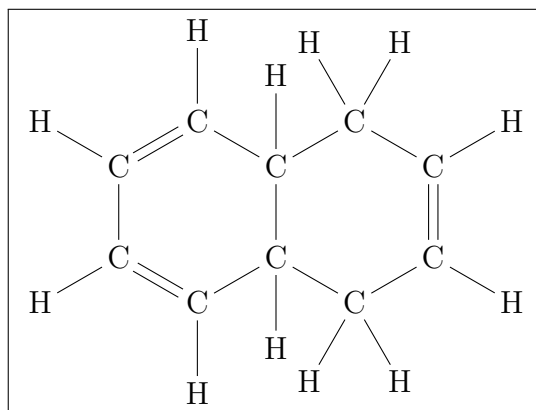


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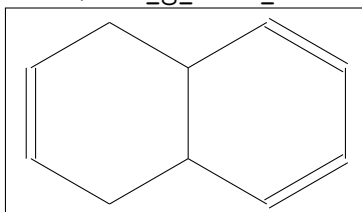


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0.3.20 $p_{\{0,18\}}$



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